

FIRST MID TERM TEST - 2026

ACCOUNTANCY

Time Allowed : 1.30 Hours]

PART - I

[Max. Marks : 45

10x1=10

Note : Choose the correct answer.

1. The root of financial accounting system is _____.
 (a) Social accounting (b) Stewardship accounting
 (c) Management accounting (d) Responsibility accounting
2. Financial position of a business is ascertained on the basis of
 (a) Journal (b) Trial balance (c) Balance sheet (d) Ledger
3. The concept which assumes that a business will last indefinitely is
 (a) Business Entity (b) Going concern (c) Periodicity (d) Conservation
4. In India, Accounting Standards are issued by _____.
 (a) Reserve Bank of India (b) The Cost and Management Accountants of India
 (c) Supreme Court of India (d) The Institute of Chartered Accountants of India
5. Real account deals with _____.
 (a) Individual persons (b) Expenses and losses (c) Assets (d) Incomes and gains
6. Withdrawal of cash from business by the proprietor should be credited to
 (a) Drawings A/c (b) Cash A/c (c) Capital A/c (d) Purchases A/c
7. The process of transferring the debit and credit items from journal to ledger accounts is called
 (a) Casting (b) Posting (c) Journalising (d) Balancing
8. If the total of the debit side of an account exceeds the total of its credit side, it means
 (a) Credit balance (b) Debit balance (c) Nil balance (d) Debit and credit balance
9. Trial balance is a _____.
 (a) Statement (b) Account (c) Ledger (d) Journal
10. The difference of totals of both debit and credit side of trial balance is transferred to:
 (a) Trading account (b) Difference account (c) Suspense account (d) Miscellaneous account

PART - II

II. Answer any Four questions. Question no.16 is compulsory.

4x2=8

11. List any two functions of Accounting.

12. Define Book keeping.

13. Write the narration with reference to the following journal entries.

In the books of Shyam

Date	Particulars	L.F.	Debit ₹	Credit ₹
2018 April 1	Cash A/c Dr. To Shyam's capital A/c (.....)		3,00,000	3,00,000
2.	Bank A/c Dr. To Cash A/c (.....)		2,70,000	2,70,000

14. Classify the following into personal, real and nominal accounts.
 (a) Capital (b) Building (c) Carriage Inwards (d) Cash
15. What are Source Documents?
16. What are the methods of preparation of Trial Balance.

PART - III

III. Answer any Four questions. Question No.22 is compulsory.

4x3=12

17. Discuss briefly to branches of Accounting.

18. "Business units last indefinitely". Mention and explain the concept on which the statement is based.

19. Give the golden rules of double entry accounting system.

20. Complete the accounting equation.

- | | | | | |
|------------|---|-----------|---|-------------|
| (a) Assets | = | Capital | + | Liabilities |
| ₹1,00,000 | = | ₹80,000 | + | ? |
| (b) Assets | = | Capital | + | Liabilities |
| ₹2,00,000 | = | ? | + | ₹40,000 |
| (c) Assets | = | Capital | + | Creditors |
| ? | = | ₹1,60,000 | + | ₹80,000 |

21. Prepare Furniture A/c from the following transactions.

2016		₹
Jan 1	Furniture in hand	2,000
1	Purchased furniture for cash	4,000
30	Sold furniture	400

22. State whether the balance of the following accounts should be placed in the debit or the credit column of the trial balance: i) Carriage outwards ii) Carriage inwards iii) Sales
iv) Purchases v) Bad debts vi) Interest paid

PART - IV

IV. Answer all Questions.

3x5=15

23. a) Show the accounting equation on the basis of the following transactions for Rani, who is dealing in automobiles.

(i)	Started business with cash	₹ 80,000
(ii)	Goods bought on credit from Ramesh	₹ 10,000
(iii)	Purchased furniture for cash	₹ 8,000
(iv)	Paid creditors by cash	₹ 8,000
(v)	Purchased goods for cash	₹ 2,000
(vi)	Paid rent by cash	₹ 500

(OR)

b) Distinguish between Book keeping and Accounting.

24. a) Jeyaseeli is a sole proprietor having a provisions store. Following are the transactions during the month of January, 2018. Journalise them.

Jan		₹
1	Commenced business with cash	80,000
2	Deposited cash with bank	40,000
3	Purchased goods by paying cash	5,000
4	Purchased goods from Lipton & Co. on credit	10,000
5	Sold goods to Joy and received cash	11,000
6	Paid salaries by cash	5,000
7	Paid Lipton & Co. by cheque for the purchases made on 4th Jan.	
8	Bought furniture by cash	4,000
9	Paid electricity charges by cash	1,000
10	Bank paid insurance premium as per standing instructions	300

(OR)

b) Pass journal entries for the following transactions and post them to ledger.

2017		₹
Aug. 1	Dharma started business with cash	70,000
8	Cash received from Ganesan	10,000
10	Rent paid	3000
20	Received commission from Anand	5000

25. a) The following balances are extracted from the books of Ravichandran on 31st December, 2016. Prepare the trial balance.

	₹		₹
Capital	1,50,000	Sales	75,000
Debtors	22,800	Return inwards	1,000
Rent received	500	Discount allowed	800
Bank overdraft	3,100	Discount received	1,000
Creditors	5,500	Wages	2,900
Premises	1,48,000	Salaries	3,500
Opening stock	10,000	Commission paid	1,100
Purchases	45,000	General expenses	2,000

(OR)

b) Distinguish between Journal and Ledger.

Standard XI ACCOUNTANCY

Time : 1.30 hrs

Part - I

Marks : 50

10×1=10

I. Choose the correct answer.

1. Which one of the following is not a branch of accounting?
 - a) Financial accounting
 - b) Management accounting
 - c) Human resource accounting
 - d) None of the above
2. Who is considered to be the internal user of the financial information?
 - a) Creditor
 - b) Employee
 - c) Customer
 - d) Government
3. The business is liable to the proprietor of the business in respect of capital introduced by the person according to _____.
 - a) Money measurement concept
 - b) Cost concept
 - c) Business entity concept
 - d) Dual aspect concept
4. GAAP's are _____.
 - a) Generally Accepted Accounting Policies
 - b) Generally Accepted Accounting Principles
 - c) Generally Accepted Accounting Provisions
 - d) None of these
5. A firm has assets of ₹1,00,000 and the external liabilities of ₹60,000. Its capital would be _____.
 - a) ₹1,60,000
 - b) ₹60,000
 - c) ₹1,00,000
 - d) ₹40,000
6. Real account deals with _____.
 - a) Individual persons
 - b) Expenses and losses
 - c) Assets
 - d) Incomes and gains
7. In double entry system of book keeping every business transaction affects _____.
 - a) minimum of two accounts
 - b) same account on two different dates
 - c) two sides of the same account
 - d) minimum three accounts
8. The process of transferring the debit and credit items from Journal to Ledger accounts is called _____.
 - a) Casting
 - b) Posting
 - c) Journalising
 - d) Balancing

9. J.F means _____.

- a) Ledger page number
c) Voucher number

- b) Journal page number
d) order number

10. The root of financial accounting system is _____.

- a) Social accounting
c) Management accounting
- b) Stewardship accounting
d) Responsibility accounting

Part - II

II. Answer any five questions. (Q.No.17 is compulsory)

5x2=10

11. Define accounting.

12. What is meant by accounting concepts?

13. What is meant by Journalising?

14. What is debit balance?

15. Classify the following into personal real and nominal accounts.

- a) Capital b) Furniture c) Salary received d) Purchases

16. Give journal entries for the following :

2025 Jan

1. Commenced business with cash ₹ 80,000
2. Deposited cash with bank ₹ 40,000

17. Fill in the missing information in the following journal entries :

Date	Particulars	L.F.	Debit ₹	Credit ₹
2018 April 1	Purchases A/c Dr. To A/c (Cash purchases)		5000	5000
April 2	 A/c Dr. To Cash a/c (Electricity charges paid)		1000	1000

Part - III

III. Answer any five questions. (Q.No.24 is compulsory)

5x3=15

18. Discuss briefly the branches of accounting.

19. "Only monetary transactions are recorded in accounting" - Explain the statement.

20. Give the golden rules of double entry accounting system.

3

21. Complete the accounting equation :-

	Assets	=	Capital	+	Liabilities
a)	₹1,00,000	=	₹80,000	+	?
b)	₹2,00,000	=	?	+	₹40,000
c)	?	=	₹30,000	+	₹45,000

22. Enter the following transactions in the Journal of Manohar who is dealing in textiles:

2018 March

1. Manohar started business with cash 60,000
2. Sold goods for cash 28,000
3. Withdrew cash for personal use 4,000

23. The following balances appeared in the books of Vinoth on Jan 1, 2018

ASSETS: Cash ₹40,000; Stock ₹50,000; Amount due from Ram ₹20,000
Machinery ₹40,000

LIABILITIES: Amount due to Vijay ₹10,000

Pass the opening journal entry.

24. What is Ledger? Explain its utilities.

Part - IV

IV. Answer all the questions.

3x5=15

25. a) Pass journal entries in the books of Sasikumar who is dealing in automobiles:

2017 Oct.

1. Commenced business with cash 40,000
2. Purchased goods from Arul on credit 70,000
3. Returned goods to Arul 10,000
4. Sold goods to Chandar for cash 30,000
5. Paid salaries through ECS 2,000

(OR)

b) Distinguish between Journal and Ledger. (Any five)

26. a) Journalise the following transactions and post them to Ledger.

2016 Jan

1. Started business with cash 10,000
5. Paid into bank 5,000
7. Purchased goods from Ram for cash 1,000

(OR)

b) Discuss the role of an accountant in the modern business world.

4

27. a) Show the accounting equation on the basis of the following transactions for Rani, who is dealing in automobiles :

	₹
i) Started business with cash	80,000
ii) Goods bought on credit from Ramesh	10,000
iii) Purchased furniture for cash	6,000
iv) Paid creditors by cash	8,000
v) Purchased goods for cash	2,000

(OR)

b) Raja has a hotel. The following transactions took place in his business. Journalise them.

Jan	₹
1. Started business with cash	3,00,000
2. Purchased goods from Rajiv on credit	1,00,000
3. Cash deposited with the bank	2,00,000
4. Borrowed loan from bank	1,00,000
5. Withdrew from bank for personal use	800



BLESSING

TUITION & TUTORIAL CENTRE



11TH STD ACCOUNTANCY

1ST MID TERM IMPORTANT QUESTIONS 2026-27

★ 2, 3 & 5 MARKS IMPORTANT QUESTIONS ★

11th Accountancy

	2M	3M	5M
Ch-1	<u>VSA</u> 1, 3, 5	<u>SA</u> 1, 4	<u>SA</u> 2, 3
Ch-2	1, 2, 3, 5	SA - 2 & 4	—
Ch-3	Pg: 40 Tee - 5 1, 3, 4, 5, 6	<u>VSA</u> 8	<u>SA</u> 2, 3, 6
			2, 3, 4, 7, 8, 10, 12

Sums - 1

UNIT - 4 LEDGER



2 MARKS

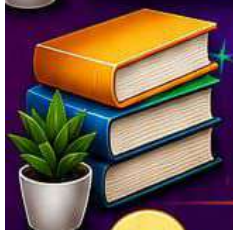
- 1, 3, 4, 5

3 MARKS

- 1

5 MARKS

- Ex - 1, 2, 3, 5, 6, 8, 10, 11
- Illu - 1, 3, 5, 6



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Standard XI CHEMISTRY

Time: 1.30 hrs.

Part – I

Marks: 50

10 x 1 = 10

I. Choose the correct answer:

1. Relative molecular mass of Hydrogen molecule is
a) 1.008U b) 18U c) 2.016U d) 16U
2. Compressibility factor 'Z' for Ideal gas at all temperature and pressure is equal to
a) Zero b) Two c) one d) Three
3. Lassaigne's test for the detection of nitrogen fails in
a) $C_6H_5-NH-NH_2 \cdot HCl$ b) $H_2N-CO-NH \cdot NH_2 \cdot HCl$
c) $NH_2-NH_2 \cdot HCl$ d) $C_6H_5 CONH_2$
4. If temperature and volume of an ideal gas is increased to twice its values, the initial pressure P becomes
a) 4P b) 2P c) P d) 3P
5. Oxidation number of Oxygen in KO_2 :
a) +2 b) -2 c) 0 d) $-\frac{1}{2}$
6. The equivalent mass of potassium permanganate in alkaline medium is
($MnO_4^- + 2H_2O + 3e^- \rightarrow MnO_2 + 4OH^-$)
a) 31.6 b) 52.7 c) 79 d) None of these
7. Which of following method used to determines amount of sulphur in organic compound?
a) Kjeldahl method b) Nitro meter c) Dumas method d) Carius method
8. Maximum deviation from ideal gas is expected from
a) $CH_{4(g)}$ b) $NH_{3(g)}$ c) $H_{2(g)}$ d) $N_{2(g)}$
9. Example for metamerism:
a) Isopentane & pentane b) Pent-1-ene & pent-2-ene
c) Propanal & propanone d) Methoxypropane & ethoxyethane
10. Select the molecule which has only one π bond
a) $CH_3-CH=CH-CH_3$ b) $CH_3-CH=CH-CHO$
c) $CH_3CH=CH-COOH$ d) All of these

Part - II

II. Answer any 5 questions. (Q.No.18 is compulsory)

5 x 2 = 10

11. What is meant by limiting reagents?
12. What do you understand by the term Mole ?
13. State Dalton Law of partial pressures.
14. What is Inversion temperature?
15. Define Avogadro hypothesis.

16. Distinguish between diffusion and effusion.
17. Give the general characteristics of organic compounds.
18. Give the structure for the following compound:
 - a) 3-methylpentane b) 2-methyl-2-ol

Part - III**III. Answer any 5 questions. (Q.No.26 is compulsory)****5 x 3 = 15**

19. Distinguish between oxidation and reduction.
20. Calculate the molar mass of the following compounds.
 - i) $[\text{CO}(\text{NH}_2)_2]$ ii) $[\text{CH}_3\text{COCH}_3]$ iii) $[\text{H}_3\text{BO}_3]$
21. What is Boyle's temperature? What happens to real gases above and below the Boyle's temperature?
22. In what way real gases differ from ideal gases.
23. Give the principle involved in the estimation of halogen in an organic compound by carius method.
24. Write a note on homologous series.
25. Explain paper chromatography.
26. Calculate the oxidation number of underlined elements.
 - a) $\underline{\text{C}}\text{O}_2$ b) $\text{H}_2\underline{\text{S}}\text{O}_4$ c) $\text{K}\underline{\text{Cl}}\text{O}_3$

Part - IV**IV. Answer all the questions.****3 x 5 = 15**

27. a) Calculate the empirical and molecular formula of a compound containing 76.6% carbon, 6.38% hydrogen and rest oxygen its vapour density is 47.

(OR)

- b) i) Define basicity. Find the basicity of ortho-phosphoric acid
 - ii) What is the empirical formula of the following?
 - a) Glucose ($\text{C}_6\text{H}_{12}\text{O}_6$) b) Acetic acid (CH_3COOH)
 28. a) i) Derive the ideal gas equation
 - ii) What is Compressibility factor?
- (OR)**
- b) Deduce the relationship between critical constants and Vanderwaal's constants
 29. a) i) Briefly explain geometrical isomerism in alkene by considering 2-butene as an example
 - ii) Explain various types of constitutional isomerism (structural isomerism) in organic compounds

(OR)

- b) Describe the classification of organic compounds based on their structure.

Stand- 11

CHEMISTRY

Time: 1.30

Part A

Choose the correct answer:

- The equivalent mass of a trivalent metal element is 9 g eq^{-1} the molar mass of its anhydrous oxide is
a) 102g b) 27g c) 270g d) 78g
- 7.5g of a gas occupies a volume of 5.6 litres at 0°C and 1 atm pressure. The gas is
a) NO b) N_2O c) CO d) CO_2
- Which one of the following is used as a standard for atomic mass?
a) ${}_6\text{C}^{12}$ b) ${}_7\text{C}^{12}$ c) ${}_6\text{C}^{13}$ d) ${}_6\text{C}^{14}$
- Splitting of spectral lines in an electric field is called
a) Zeeman effect b) Shielding effect
c) Compton effect d) Stark effect
- The total number of orbitals associated with the principal quantum number $n=3$ is
a) 9 b) 8 c) 5 d) 7
- The value of the gas constant R is
a) $0.082 \text{ dm}^3 \text{ atm}$ b) $0.987 \text{ cal mol}^{-1} \text{ K}^{-1}$
c) $8.3 \text{ J mol}^{-1} \text{ K}^{-1}$ d) $8 \text{ erg mol}^{-1} \text{ K}^{-1}$
- What is the density of N_2 gas at 227°C and 5.00 atm pressure?
a) 1.40 g/L b) 2.81 g/L c) 3.41 g/L d) 0.29 g/L
- The IUPAC name of the compound $\text{CH}_3-\text{CH}=\text{C}-\text{CH}_2-\text{CH}_3$ is

$\begin{array}{c} | \\ \text{CH}_2-\text{CH}_2-\text{CH}_3 \end{array}$

a) 3-Ethyl-2-hexene b) 3-propyl-3-hexene
c) 4-Ethyl-4-hexene d) 3-propyl-2-hexene
- Nitrogen detection in an organic compound is carried out by Lassaigne's test. The blue colour formed is due to the formation of
a) $\text{Fe}_3[\text{Fe}(\text{CN})_6]_2$ b) $\text{Fe}_4[\text{Fe}(\text{CN})_6]_3$
c) $\text{Fe}_4[\text{Fe}(\text{CN})_6]_2$ d) $\text{Fe}_3[\text{Fe}(\text{CN})_6]_3$
- Ortho and Para-nitro phenol can be separated by
a) azeotropic distillation b) destructive distillation
c) steam distillation d) cannot be separated

Part - B

Answer the following questions. Q No.

Define: Equivalent mass

State Dalton's exclusion

13) Define: Dalton's

- 14) What is Functional group?
- 15) Can a Vander Waals gas with $a = 0$ be liquefied? Explain.
- 16) What is sublimation? Give example.
- 17) Give the electronic configuration of Mn^{2+} and Cr^{3+} .

Part - C**III. Answer any five questions. Q.No. 24 is compulsory:****5×3=15**

- 18) What do you understand by the term oxidation number?
- 19) What are the n and l values for $3p_x$ and $4d_{x^2-y^2}$ electron?
- 20) Derive ideal gas equation.
- 21) Give the general characteristics of organic compounds.
- 22) State Boyle's law.
- 23) What is optical isomerism? With example.
- 24) How many moles of ethane is required to produce 44g of CO_2 after combustion?

Part - D**IV. Answer all the questions:****3×5=15**

- 25) a) Derive the values of critical constants in terms of Vander Waals constant.

(OR)

- b) i) What is Limiting reagents?
- ii) Distinguish between oxidation and reduction.

- 26) a) Describe the classification of organic compounds based on their structure.

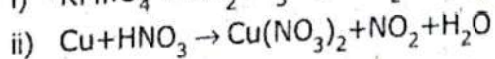
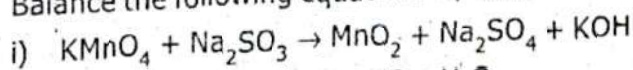
(OR)

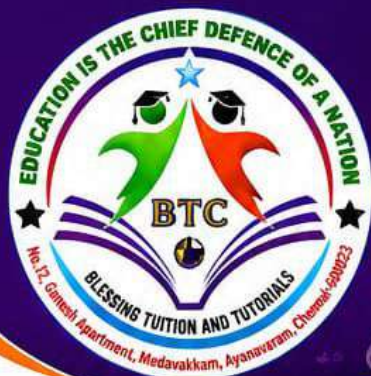
- b) i) Determine the values of all the four quantum numbers of the 8th electron in O - atom and 15th electron in Cr atom and the last electron in chromium.
- ii) Describe the Aufbau principle.

- 27) a) Explain various types of structural isomerism in organic compounds.

(OR)

- b) Balance the following equations by oxidation number method.





11TH STD CHEMISTRY

1ST MID TERM IMPORTANT QUESTION

2026 & 27



UNIT – 1 Basic Concepts of Chemistry and Chemical Calculations

2 MARKS

26, 27, 28, 29, 30, 31, 37, 38, 40

Example – Empirical Formula (1,2)

5 MARKS

42, 43

- 1.5.1 - Equivalent Mass Calculation
- 1.7.2 - Limiting Reagents
- 1.8.1 - oxidation number Calculation

UNIT – 2 Quantum Mechanical Model of Atom



2 MARKS

26, 27, 32, 39

- 2.1.1 – Bohr atom model
- 2.1.2 – Limitation of Bohr's atom model

5 MARKS

31, 33, 40

- 2.2.1 – de Broglie eq
- 2.3 – Heisenberg's uncertainty principle
- 2.5 – Quantum numbers and types

UNIT – 6 Gaseous States



2 MARKS

26, 27, 28, 29, 30, 32*(a), 34, 37, 38

- Charles Law, Boyle's Law, Gay-Lussac's Law, Avogadro Law, Dalton's law, Graham's Law

5 MARKS

40, 41

- 6.3 - Ideal gas equation

UNIT – 7 Thermodynamics



2 MARKS

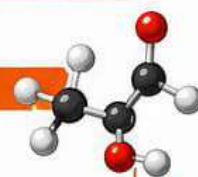
26, 27, 29, 30, 32, 33, 34, 36, 37, 38, 39, 42

- Relation between C_p and C_v

5 MARKS

43, 45, 46, 47, 50, 51, 52

UNIT – 11 Fundamentals Of Organic Chemistry



2 MARKS

31, 33, 35, 42, 43, 46

- 38 – IUPAC (i to vi)
- 39 – (i to x)

5 MARKS

32, 40, 41, 45



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